# Photoreactions of 2-(N-Alkylarylamino)succinimides and Related Compounds

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Abstract: On irradiation (254 nm), 2-(N-methylanilino)-N-phenylsuccinimide (6a) underwent homolysis of the C(2)-N bond to yield fission products N-methylaniline and N-phenylsuccinimide (12), along with the products of ortho- and para-rearrangement 2-[2-(N-methylamino)phenyl]-N-phenylsuccinimide (13) and 2-[4-(N-methylamino)phenyl]-N-phenylsuccinimide (14a) respectively. Similarly, irradiation of 2-(N-methylanilino)succinimide (6b) and N-benzyl-2-(N-methylanilino)succinimide (6c) gave N-methylaniline and the corresponding products of para-rearrangement. In contrast, both fission and type II cyclisation photoproducts 12 and 17 respectively were formed from N-phenyl-2-(1,2,3,4-tetrahydroquinolin-1-yl)succinimide (16) and only the type II cyclisation product 1-hydroxy-2,7-diphenyl-2,7-diazabicyclo-[3.3.0]octan-3-one (21) was obtained from 2-(N-methylanilinomethyl)-N-phenylsuccinimide (20).

The photochemistry of N-alkylsuccinimides and N-alkylphthalimides has been studied extensively during recent years. One of the most common photoreactions is abstraction of hydrogen from the  $\gamma$ -position of the N-alkyl group, by one of the carbonyl oxygens, with subsequent cyclisation of the resulting biradical. In the case of succinimides 1, this leads to a product 2 containing an unstable 2-hydroxyazetidine ring, and rearrangement to a stable azepine-2,5-dione follows. Similar photoreactions occur with N-alkylphthalimides. In view of these photoreactions and also of the successful photocyclisation of N-alkylarylaminomethyl aryl ketones (e.g. 4, R = aryl) to azetidinols 5,3 we decided to investigate the photochemistry of 2-(N-methylanilino)succinimides 6 and related succinimides. A hydrogen abstraction-cyclisation sequence (type II cyclisation) similar to the reaction  $1 \rightarrow 2$  would lead to interesting functionalised azetidines (e.g. 8).

# **RESULTS**

A series of 2-substituted succinimides, contained in quartz tubes, were irradiated using 254 nm lamps (see Experimental). The results are given in the Table. Irradiation of a 2% solution of succinimide 6a in benzene/methanol yielded a complex mixture from which N-methylaniline (11) and a fraction, presumably consisting of N-phenylsuccinimide (12) and the product 14a of para-rearrangement, were isolated. Treatment of this fraction with acetic anhydride yielded N-phenylsuccinimide and the N-acetyl derivative 15 of amine 14a. When a more dilute solution (0.6%) of succinimide 6a was irradiated, the product 13 of ortho-rearrangement was isolated, along with photoproducts 11 and 14a, from the reaction mixture. Irradiation of the succinimides 6b (in methanol) and 6c (in benzene/methanol) also yielded N-methylaniline and the corresponding para- rearrangement products 14b or 14c respectively. In contrast, irradiation of the tetrahydroquinolinyl-succinimide 16 yielded the type II cyclisation product 17 as well as the fission product 12. No significant products could be isolated from the complex photolysis mixtures obtained from the succinimide 18 and the maleimide 19. Cyclisation to the 2,7-diazabicyclo-octanol 21 occurred in good yield on irradiation of the N-methylanilinomethyl-succinimide 20, and fission photoproducts 12 and 23 were obtained from the tolylthio-succinimide 22.

Preparation of succinimides 6, 16, 18, 20 and 22 and maleimide 19. - Succinimides 16 and 18 were obtained from the reaction of N-phenylmaleimide (24) with the appropriate secondary amine and succinimide 6c was produced in a similar manner from N-benzylmaleimide and N-methylaniline.

Treatment of 2-bromo-N-phenylmaleimide (25) with N-methylaniline in the presence of triethylamine yielded the maleimide 19. Succinimide 20 was formed by addition of N-methylaniline to 2-methylene-N-phenylsuccinimide (26), the product of cyclisation of itaconic acid mono-anilide (27). Succinimides 6a, 6b, and 22 were prepared using literature procedures (see Experimental).

Table: Irradiation of 2-Substituted Succinimides.a

Succinimide	Solvent	Time (h)	Products (Yield %)b
ба	PhH/MeOH (1:1) <sup>c</sup>	15	11 (23), 12 (12), 15 (10) <sup>d</sup> 6a (45)
6a	PhH/MeOH (1:1) <sup>e</sup>	15	11 (9), 13 (20), 14a (9), 6a (10)
6b	МеОН	12	11 (12), 14b (20), 6b (10)
6с	PhH/MeOH (1:1)	10	11 (20), 14c (16), 6c (11)
16	CHCl <sub>3</sub>	8	<b>12</b> (36), <b>17</b> (20), <b>16</b> (41)
20	MeOH	21	<b>21</b> (80), <b>20</b> (50)
22	PhH	6	<b>12</b> (53), <b>23</b> (11), <b>22</b> (27)

<sup>&</sup>lt;sup>a</sup> Irradiations were carried out on solutions contained in quartz tubes using a Rayonet RPR - 100 photoreactor fitted with 254 nm lamps.

b Yields are based on the amount of starting material consumed.

c 2% solution.
d After treatment with acetic anhydride.
0.67% solution.

# DISCUSSION

Unlike N-alkylsuccinimides 1, which undergo type II cyclisation in moderate yields, the 2-N-methylanilinosuccinimides 6a - 6c cleave to radicals 9 and 10 on irradiation. In each case the fission product N-methylaniline (11) and radical-recombination product 14a - 14c respectively were formed. Only in the case of succinimide 16 was a type II cyclisation product, 17, isolated from the reaction mixture. The lack of type II cyclisation by succinimides 6 probably reflects the relatively high strain in the transition state leading to biradical 7 or to product 8, or both. This could result in C(2)-N bond cleavage, to radicals 9 and 10, being the more favoured reaction pathway. This cleavage to radicals followed by recombination to give products of ortho- and/or para-rearrangement is similar to the photorearrangement<sup>4</sup> of aminoketones 4 (R = alkyl), which, unlike the corresponding aryl aminoketones 4 (R = aryl), do not undergo type II cyclisation.

In contrast to succinimides 6, the tetrahydroquinolinyl-succinimide 16 undergoes both type II cyclisation to the azetidinol 17 as well as C(2)-N bond cleavage. In imide 16, one of the two C(2') hydrogens is favourably placed for transfer to the adjacent carbonyl group and hence only rotation about the C(2)-N bond is required to bring the two interacting hydrogen and oxygen atoms near enough to each other for type II cyclisation to occur. In the case of succinimides 6, rotation about both the C(2)-N and N-CH<sub>3</sub> bonds is required to achieve the conformation necessary for type II cyclisation and hence this reaction will be less favoured for these imides than it is for succinimide 16. In succinimide 20, strain in the transition states leading to product 21 is unlikely to be severe and photocyclisation to a 5-membered ring occurs readily. A related  $\delta$ -hydrogen abstraction followed by cyclisation to a 5-membered ring was reported for the *N*-alkylsuccinimides 28, which lack the C( $\gamma$ ) hydrogen necessary for type II cyclisation to a 4-membered ring.<sup>5</sup>

**28** R = Me or R,R =  $(CH_2)_n$  (n = 4, 5, or 6)

# **EXPERIMENTAL**

I.r. spectra were recorded as Nujol mulls (for solids) or as thin films (for oils) and n.m.r. spectra were recorded at 300 MHz with CDCl<sub>3</sub> as solvent unless otherwise stated. Electron impact mass spectral data are recorded except where the use of chemical ionisation (C.I.) is noted. 'Chromatography' refers to 'flash' column chromatography over Merck Kieselgel 60H and light petroleum had b.p. 60-80°C unless otherwise stated. The following compounds were prepared according to literature procedures: 2-(*N*-methylanilino)-*N*-phenylsuccinimide (6a),<sup>6</sup> 2-(*N*-methylanilino)succinimide (6b),<sup>7</sup> *N*-phenyl-2-(*p*-tolylthio)succinimide (22),<sup>8</sup> *N*-benzylmaleimide,<sup>9</sup> *N*-phenylmaleimide (24),<sup>10</sup> 2-bromo-*N*-phenylmaleimide (25),<sup>11</sup> itaconic acid mono-anilide (27),<sup>12</sup> *N*-phenylsuccinimide (12),<sup>13</sup> and di-*p*-tolyl disulphide (23).<sup>14</sup>

N-Benzyl-2-(N-methylanilino)succinimide (6c). - A solution of N-methylaniline (0.43 g) and N-benzylmaleimide (0.75 g) in glacial acetic acid (8 ml) was heated for 9 h on the water-bath. The resulting solid was collected and crystallised from light petroleum (b.p. 80-100°C) to give N-benzyl-2-(N-methylanilino)succinimide (0.58 g, 49%), m.p. 93°C (Found: C, 73.2; H, 6.1; N, 9.8. C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub> requires C, 73.4; H, 6.2; N, 9.5%);  $v_{\text{max}}$  1770 and 1695 cm<sup>-1</sup>;  $\delta$  2.66 [1H, dd, J 5.6 and 18.5 Hz, C(3)-H], 2.74 (3H, s, NMe), 3.0 [1H, dd, J 9.0 and 18.5 Hz, C(3)-H], 4.69 and 4.71 (2 x 1H, doublets J 13.9 Hz, NCH<sub>2</sub>), 4.80 [1H, dd, J 5.6 and 9.0 Hz, C(2)-H], 6.73-6.83 (3H, m, ArH), and 7.17-7.40 (7H, m, ArH); m/z 294 ( $M^+$ , 71%), 203 (18), 189 (14), 160 (39), 133 (61), 106 (100), 91 (79), 77 (51), 65 (17), and 51 (21).

N-Phenyl-2-(1,2,3,4-tetrahydroquinolin-1-yl)succinimide (16). - A solution of N-phenyl-maleimide (1.93 g) and 1,2,3,4-tetrahydroquinoline (1.49 g) in glacial acetic acid (15 ml) was kept for 18 h at room temperature. The resulting precipitate was collected and crystallised from light petroleum (b.p. 80-100°C) to give N-phenyl-2-(1,2,3,4-tetrahydroquinolin-1-yl)succinimide (2.7 g, 79%), m.p. 195-197°C (Found: C, 74.5; H, 5.8; N, 9.0. C<sub>19</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub> requires C, 74.5; H, 5.9; N, 9.1%); ν<sub>max</sub>. 1765 and 1680 cm<sup>-1</sup>; δ (90 MHz) 1.8-2.2 (2H, m, CH<sub>2</sub>), 2.6-3.46 (6H, m, 3 x CH<sub>2</sub>), 5.18 [1H, dd, J 6.5 and 9.0 Hz, C(2)-H], and 6.5-7.7 (9H, m, ArH).

2-(Indolin-1-yl)-N-phenylsuccinimide (18). - Indoline (1.9 g) was allowed to react with N-phenylmaleimide (2.56 g), using the procedure described in the preceding experiment, to give 2-(indolin-1-yl)-N-phenylsuccinimide (2.57 g, 59%), m.p. 118-120°C (from toluene) (Found: C, 73.9; H, 5.4; N, 9.4.  $C_{18}H_{16}N_2O_2$  requires C, 74.0; H, 5.5; N, 9.6%);  $v_{max}$ . 1780 and 1700 cm<sup>-1</sup>;  $\delta$  (90 MHz) 2.7-3.2 (4H, m, 2 x CH<sub>2</sub>), 3.25-3.6 (2H, m, NCH<sub>2</sub>), 4.88 [1H, dd J 6.5 and 9.0 Hz, C(2)-H], and 6.3-7.6 (9H, m, ArH).

2-(N-Methylanilino)-N-phenylmaleimide (19). A solution of 2-bromo-N-phenylmaleimide (0.504 g), N-methylaniline (0.214 g), and triethylamine (0.35 ml) in dry xylene (5 ml) was heated under reflux for 1 h. The cooled mixture was extracted into ether and the ethereal extract was washed with water, dried, and evaporated. Chromatographic separation of the residual oil, eluting with 1:1 dichloromethane/light petroleum, yielded 2-(N-methylanilino)-N-phenylmaleimide (0.33 g, 60%) as a yellow oil (Found:  $M^+$  278.1055.  $C_{17}H_{14}N_2O_2$  requires  $M^+$  278.1055);  $v_{max}$  1760, 1712, and 1620 cm<sup>-1</sup>;  $\delta$  3.56 (3H, s, NMe), 5.13 [1H, s, C(3)-H], and 7.28-7.44 (10H, m, ArH); m/z 278 ( $M^+$ , 100%), 249 (21), 158 (18), 131 (83), 116 (58), 106 (19), 91 (26), 77 (69), 65 (14), and 51 (38).

2-(N-Methylanilinomethyl)-N-phenylsuccinimide (20). - (i) A mixture of itaconic acid monoanilide (5 g), acetic anhydride (11 ml), and anhydrous sodium acetate (1.0 g) was heated on the water-bath until all the anilide had dissolved. The mixture was cooled, extracted into ether, and the extract was washed with 10% aqueous sodium hydroxide and then dried. After evaporation, the residue was chromatographed, eluting with 1:1 dichloromethane/light petroleum, to give 2-methylene-N-phenyl-succinimide (26) (2.7 g, 59%), m.p. 120°C (Found: C, 70.3; H, 4.9; N, 7.4. C<sub>11</sub>H<sub>9</sub>NO<sub>2</sub> requires C,

70.6; H, 4.8; N, 7.5%);  $v_{\text{max}}$  1711 and 1665 cm<sup>-1</sup>;  $\delta$  3.5 (2H, apparent t J 2.3 Hz, CH<sub>2</sub>), 5.73 (1H, apparent t J 2.0 Hz, C=CH), 6.46 (1H, apparent t J 2.5 Hz, C=CH), and 7.24-7.50 (5H, m, Ph); m/z (C.I.) 188 [(M + H)<sup>+</sup>, 12%], 187 (M<sup>+</sup>, 100), 159 (6), 130 (16), 119 (15), 91 (26), 77 (13), 68 (93), and 64 (27).

(ii) A solution of 2-methylene-N-phenylsuccinimide (2.0 g) and N-methylaniline (1.25 g) in glacial acetic acid (20 ml) was heated under reflux for 18 h. The mixture was cooled, extracted into ether, and the ethereal extract was washed with 10% aqueous sodium hydroxide, dried, and evaporated. Chromatographic separation, eluting with 3:1 light petroleum/ethyl acetate, gave 2-(N-methylanilino-methyl)-N-phenylsuccinimide (1.1 g, 35%) as an oil (Found: C, 72.9; H, 6.4; N, 9.1.  $C_{18}H_{18}N_2O_2$  requires C, 73.4; H, 6.2; N, 9.5%);  $v_{max}$ . 1777 and 1712 cm<sup>-1</sup>;  $\delta$  2.76 (1H, dd, J 5.3 and 18.4 Hz) and 2.92 (1H, dd, J 9.1 and 18.4 Hz) [C(3)H<sub>2</sub>], 3.0 (3H, s, NMe), 3.30-3.36 [1H, m, C(2)-H], 3.68 (1H, dd J 9.1 and 15.0 Hz) and 4.0 (1H, dd J 5.0 and 15.0 Hz) (NCH<sub>2</sub>), and 6.75-6.80 and 7.22-7.49 (10H, multiplets, ArH); m/z 294 ( $M^+$ , 6%), 120 (100), 104 (7), 91 (5), 77 (13), and 55 (6).

# Irradiation of 2-Substituted Succinimides

A solution of the succinimide contained in a quartz tube was irradiated in a Rayonet RPR-100 photoreactor fitted with 254 nm lamps. The solvent was evaporated and the products were generally isolated by 'flash' column chromatography over silica gel.

Irradiation of 2-(N-methylanilino)-N-phenylsuccinimide (6a). - (i) A solution of imide 6a (1.2 g) in 1:1 benzene/methanol (60 ml) was irradiated for 15 h. Chromatography (at atmospheric pressure) of the product mixture over Hopkin & Williams silica gel M.F.C., eluting with toluene, yielded N-methylaniline (58 mg). Further elution with 19:1 toluene/ethyl acetate gave a mixture (180 mg) of N-phenylsuccinimide (12) and 2-[4-(N-methylamino)phenyl]-N-phenylsuccinimide (14a);  $v_{max}$  3400, 1770, and 1690 cm<sup>-1</sup>;  $\delta$  (product 12) 2.79 (4H, s, 2 x CH<sub>2</sub>) and 7.0-7.5 (5H, m, Ph);  $\delta$  (product 14a) 1.7 (1H, broad signal, NH), 2.78 (3H, s, NMe), 2.92 [1H, dd J 4.7 and 18.5 Hz, C(3)-H], 3.30 [1H, dd J 9.5 and 18.5 Hz, C(3)-H], 4.04 [1H, dd J 4.7 and 9.5 Hz, C(2)-H], 6.57-6.60 and 7.07-7.1 (4H, AA'BB'm, C<sub>6</sub>H<sub>4</sub>), and 7.1-7.7 (5H, m, Ph); m/z (C.I.) 298 [(M + NH<sub>4</sub>)+, 3%] and 281 [(M + H)+, 3] (M refers to product 14a).

The above mixture of 12 and 14a (160 mg) was heated under reflux for 5 min. with acetic anhydride (1 ml). Water (ca. 3 ml) was added and the mixture was heated to the boil and then cooled. Filtration yielded 2-[4-(N-acetyl-N-methylamino)phenyl]-N-phenylsuccinimide (15) (65 mg) (Found:  $M^+$  322.1318. C<sub>19</sub>H<sub>18</sub>N<sub>2</sub>O<sub>3</sub> requires  $M^+$  322.1317);  $\nu_{max}$ . 1780 and 1700 cm<sup>-1</sup>;  $\delta$  (90 MHz) 1.92 (3H, s, COMe), 3.25 (3H, s, NMe), 2.8-3.4 (2H, m, CH<sub>2</sub>), 4.25 (1H, dd J 6 and 10 Hz, CH), and 6.5-8.0 (9H, m, ArH). Evaporation of the filtrate gave N-phenylsuccinimide (12) (45 mg).

(ii) The imide  $\mathbf{6a}$  (1.0 g) in 1:1 benzene/methanol (150 ml) was irradiated for 15 h. Chromatography of the photoproduct mixture, eluting with 3:1 light petroleum/ethyl acetate, gave first N-methylaniline (30 mg) and then imide  $\mathbf{6a}$  (0.1 g). Further elution with 2:1 light petroleum/ethyl acetate yielded first 2-[4-(N-methylanilino)phenyl]-N-phenylsuccinimide (14a) (80 mg) (Found:  $M^+$  280.1221. C<sub>17</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub> requires  $M^+$  280.1212);  $V_{\text{max}}$  3410, 1767, and 1705 cm<sup>-1</sup>;  $\delta$  2.87 (3H, s, NMe), 3.13

[1H, dd J 6 and 18 Hz, C(3)-H], 3.20 [1H, dd J 9 and 18 Hz, C(3)-H], 4.45 [1H, dd J 6 and 9 Hz, C(2)-H], 6.64-6.67 and 7.57-7.60 (4H, AA'BB'm, C<sub>6</sub>H<sub>4</sub>), and 7.3-7.5 (5H, m, Ph); m/z 280 ( $M^+$ , 4%), 187 (10), 160 (42), 145 (68), 120 (14), 111 (9), 102 (100), 84 (31), 73 (67), 65 (15), and 57 (43). Further elution gave 2-[2-(N-methylamino)phenyl]-N-phenylsuccinimide (13) (0.18 g) (Found:  $M^+$  280.1223. C<sub>17</sub>H<sub>16</sub>N<sub>2</sub>O<sub>2</sub> requires  $M^+$  280.1212); v<sub>max</sub>. 3315, 1695, and 1614 cm<sup>-1</sup>;  $\delta$  2.7 [1H, dd J 6.7 and 15.7 Hz, C(3)-H], 3.03 [1H, dd J 6.7 and 15.7 Hz, C(3)-H], 3.18 (3H, s, NMe), 3.89 [1H, t J 6.7 Hz, C(2)-H], 6.80-7.55 (9H, m, ArH), and 9.13 (1H, s, NH); m/z 280 ( $M^+$ , 17%), 187 (5), 159 (100), 145 (6), 131 (14), 117 (12), 92 (22), 77 (19), and 65 (16).

Irradiation of 2-(N-Methylanilino)succinimide (**6b**). - A solution of imide **6b** (1.0 g) in methanol (70 ml) was irradiated for 12 h. Chromatography of the photoproduct mixture, eluting with 4:1 light petroleum/ethyl acetate, gave N-methylaniline (56 mg). Elution with 3:1 light petroleum/ethyl acetate gave imide **6b** (0.1 g) and elution with 2:1 light petroleum/ethyl acetate yielded a solid (0.2 g) which on crystallisation from toluene gave 2-[4-(N-methylamino)phenyl]succinimide (**14b**) (0.18 g), m.p. 150-151°C (Found: C, 64.55; H, 6.2; N, 13.7. C<sub>11</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub> requires C, 64.7; H, 5.9; N, 13.7%);  $\nu_{max}$ . 3350, 1768, and 1705 cm<sup>-1</sup>;  $\delta$  2.52 [1H, dd J 5.0 and 18.4 Hz, C(3)-H], 2.59 (3H, s, NMe), 2.94 [1H, dd J 9.6 and 18.4 Hz, C(3)-H], 3.68 [1H, dd J 5.0 and 9.6 Hz, C(2)-H], 3.96 (1H, broad s, NH), 6.36-6.39 and 6.81-6.84 (4H, AA'BB'm, C<sub>6</sub>H<sub>4</sub>), and 10.62 (1H, broad s, NHCO); m/z 204 ( $M^+$ , 48%), 133 (100), 131 (47), 91 (10), 77 (19), 66 (49), 65 (28), 53 (19), and 51 (17).

Irradiation of N-Benzyl-2-(N-methylanilino)succinimide (6c). - A solution of imide 6c (1.2 g) in 1:1 benzene/methanol (60 ml) was irradiated for 10 h. Chromatography of the photoproduct mixture, eluting with 1:1 cyclohexane/dichloromethane, gave N-methylaniline (79 mg). Further elution with 1:2 cyclohexane/dichloromethane yielded first imide 6c (0.13 g) and then an impure fraction which was rechromatographed to give N-benzyl-2-[4-(N-methylamino)phenyl]succinimide (14c) (0.17 g) (Found:  $M^+$  294.1324. C<sub>18</sub>H<sub>18</sub>N<sub>2</sub>O<sub>2</sub> requires  $M^+$  294.1368);  $v_{max}$ . 3400, 1770, and 1690 cm<sup>-1</sup>;  $\delta$  1.7 (1H, broad signal, NH), 2.74 [1H, dd J 4.7 and 18.5 Hz, C(3)-H], 2.80 (3H, s, NMe), 3.13 [1H, dd J 9.5 and 18.5 Hz, C(3)-H], 3.89 [1H, dd J 4.7 and 9.5 Hz, C(2)-H], 4.65 and 4.69 (2 x 1H, doublets J 14 Hz, CH<sub>2</sub>), 6.61-6.64 and 6.94-6.97 (4H, AA'BB'm, C<sub>6</sub>H<sub>4</sub>), and 7.22-7.37 (5H, m, Ph); m/z 294 ( $M^+$ , 59%), 161 (14), 133 (100), 132 (53), 91 (26), 77 (16), 65 (12), and 49 (13).

Irradiation of N-Phenyl-2-(1,2,3,4-tetrahydroquinolin-1-yl)succinimide (16). - A solution of imide 16 (1.4 g) in chloroform (60 ml) was irradiated for 8 h. Chromatography of the photoproduct mixture, eluting with 9:1 chloroform/ethyl acetate, yielded first imide 16 (0.573 g), then N-phenyl-succinimide (12) (171 mg), and finally the pyrrolo-azetoquinoline 17-(168 mg), m.p. 145-147°C (from methanol) (Found: C, 74.0; H, 6.2; N, 9.0.  $C_{19}H_{18}N_{2}O_{2}$  requires C, 74.5; H, 5.9; N, 9.1%);  $v_{max}$ . 3300, 1687, and 1670 cm<sup>-1</sup>;  $\delta$  (90 MHz) 1.6-2.2 (2H, m, CH<sub>2</sub>), ca. 2.6-3.1 (4H, m, ArCH<sub>2</sub> and CH<sub>2</sub>CO), ca. 3.5-4.1 (2H, m, 2 x NCH), 6.7-7.9 (9H, m, ArH), and 9.2 (1H, s, OH); m/z 306 ( $M^+$ , 80%), 213 (26), 186 (85), 185 (100), 184 (34), 172 (29), 158 (44), 156 (42), and 130 (31).

Irradiation of 2-(N-Methylanilinomethyl)-N-phenylsuccinimide (20). - A solution of imide 20 (1.0 g) in methanol (200 ml) was irradiated for 21 h. Chromatography of the photoproduct mixture, eluting with 3:1 light petroleum/ethyl acetate, gave imide 20 (0.5 g). Further elution with 1:2 light petroleum/ethyl acetate yielded 1-hydroxy-2,7-diphenyl-2,7-diazabicyclo[3.3.0]octan-3-one (21) (0.4 g), m.p. 146-147°C (from toluene) (Found: C, 73.3; H, 6.3; N, 9.2.  $C_{18}H_{18}N_{2}O_{2}$  requires C, 73.4; H, 6.2; N, 9.5%);  $v_{max}$ . 3330 and 1680 cm<sup>-1</sup>; δ 2.30-2.39 (1H, m, CH), 2.81-2.93 (2H, m, CH<sub>2</sub>), 3.15-3.2 (1H, m, CH), 3.27 and 3.53 [2 x 1H, doublets J 10.6 Hz, C(8)H<sub>2</sub>], 3.67-3.73 (1H, m, CH), 4.45 (1H, broad signal, OH), 6.43-6.51 and 6.64-7.38 (10H, multiplets, 2 x Ph); m/z (C.I.) 295 [(M + H)+, 100%], 251 (15), 120 (16), 106 (9), and 94 (39).

Irradiation of N-Phenyl-2-(p-tolylthio)succinimide (22). - A solution of imide 22 (1.0 g) in benzene (100 ml) was irradiated for 6 h. Chromatography of the photoproduct mixture, eluting with chloroform, yielded di-p-tolyl disulphide (23) (35 mg). Further elution with 9:1 chloroform/ethyl acetate gave imide 22 (0.27 g) and then N-phenylsuccinimide (12) (0.23 g).

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